

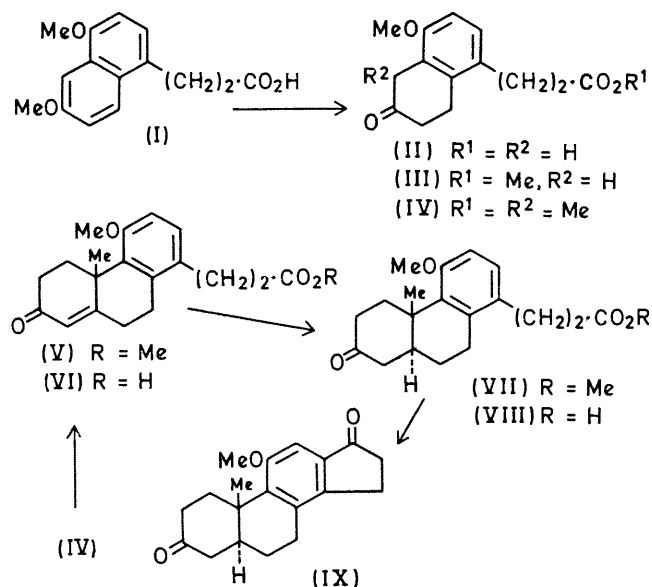
## Total Synthesis of a Ring-C-aromatic 18-Nor-steroid

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**Summary** The total synthesis of 11-methoxy-18-nor-5 $\alpha$ -androsta-8(9),11,13(14)-triene-3,17-dione is reported

IN view of the recent interest<sup>1</sup> in c-ring-aromatic steroids, we report our work on the total synthesis of such a compound



Reduction<sup>†</sup> of the known acid (I)<sup>2</sup> with sodium and alcohol in liquid ammonia, followed by mild acid hydrolysis, furnished a mixture of products. Chromatography of this mixture afforded the crystalline acid<sup>‡</sup> (II) (64%), m p 126—127°,  $\lambda_{\max}$  275 and 283 nm ( $\epsilon$  1995 and 1963),  $\nu_{\max}$  1712  $\text{cm}^{-1}$ ,  $\tau$  ( $\text{CDCl}_3$ ) 6.77—7.57 (8H), 6.45 (2H, s), 6.22 (3H, s), and 3.12 (2H, q,  $J$  8 Hz). The keto-acid (II)

furnished a methyl ester (III), m p 84°,  $\lambda_{\max}$  276 and 283 nm ( $\epsilon$  1905 and 1928),  $\nu_{\max}$  1720  $\text{cm}^{-1}$ ,  $\tau$  ( $\text{CDCl}_3$ ) 6.81—7.57 (8H), 6.46 (2H, s), 6.32 (3H, s), 6.20 (3H, s), and 3.17 (2H, q,  $J$  8.6 Hz). The enamine, m p 98—99°,  $\nu_{\max}$  1728, 1611, and 1568  $\text{cm}^{-1}$ , prepared from (III) was alkylated with methyl iodide in methanolic solution. The alkylated product, obtained as a liquid, was purified by evaporative distillation. The resulting homogeneous (t.l.c. and column chromatography) product provided a crystalline semicarbazone (91%), m p 203—204°. The structure (IV) for this product is supported from its spectral characteristics  $\lambda_{\max}$  276 and 283 nm ( $\epsilon$  1905 and 1950),  $\nu_{\max}$  (film) 1711 and 1733  $\text{cm}^{-1}$ ,  $\tau$  ( $\text{CCl}_4$ ) 8.74 (3H, d,  $J$  7.5 Hz), 7.00—7.80 (9H), 6.41 (3H, s), 6.23 (3H, s), and 3.25 (2H, q,  $J$  7.2 Hz).

The reaction of the keto-ester (IV) with the methoxide of 1-diethylaminobutan-3-one furnished a higher-boiling product which was characterised as (V) from elemental analyses and spectral properties,  $\lambda_{\max}$  229, 278, and 284 nm ( $\epsilon$  21,880, 3236, and 3199),  $\nu_{\max}$  1715 and 1660  $\text{cm}^{-1}$ ,  $\tau$  ( $\text{CCl}_4$ ) 8.32 (3H, s), 6.95—8.22 (12H), 6.37 (3H, s), 6.18 (3H, s), 4.32 (1H, s), and 3.20 (2H, q,  $J$  8.4 Hz), 2,4-dinitrophenylhydrazone, m p 182—183°,  $\lambda_{\max}$  ( $\text{CHCl}_3$ ) 390 nm ( $\epsilon$  34,040), semicarbazone, m p 239—240°. Alkaline hydrolysis of (V) produced a noncrystalline acid, characterised as (VI) by its crystalline semicarbazone, m p 243—244°. Birch reduction of (VI) followed by esterification ( $\text{CH}_2\text{N}_2$ ) furnished the crystalline methyl ester (VII), m p 89°,  $\lambda_{\max}$  275 and 282 nm ( $\epsilon$  1938 and 1945),  $\nu_{\max}$  1723 and 1705  $\text{cm}^{-1}$ . This ester, on alkaline hydrolysis, afforded the corresponding acid (VIII), m p 140°,  $\lambda_{\max}$  276 and 283 nm ( $\epsilon$  1823 and 1833),  $\nu_{\max}$  1705  $\text{cm}^{-1}$  (broad). The *trans*-structure (VIII) for the acid is assigned by analogy with the well-known stereochemical course of the Birch reduction. Polyphosphoric acid cyclisation of (VIII) furnished in excellent yield the crystalline c-aromatic compound (IX), m p 172°,  $\lambda_{\max}$  264 and 319 nm ( $\epsilon$  15,490 and 6542),  $\nu_{\max}$  1704  $\text{cm}^{-1}$  (broad).

As far as we know, this is the first synthesis of a ring-c-aromatic steroid with an angular methyl group at C-10 and an oxygen function at C-11.

<sup>†</sup> The reaction conditions for this reduction are critical, slight change in the conditions produces  $\beta$ -(1,2,3,4-tetrahydro-8-methoxy-5-naphthyl)propionic acid as the main product of the reduction.

<sup>‡</sup> All compounds described herein gave expected elemental analyses. Uv spectra were taken in ethanol solution, and ir spectra were measured in chloroform solution unless otherwise stated.

The structure recently established<sup>3</sup> for the mould metabolite, viridin, is the first naturally occurring c-aromatic steroid.

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<sup>1</sup> C. F. Hammer, D. S. Savage, J. B. Thomson, and R. Stevenson, *Tetrahedron*, 1964, **20**, 929; D. Levy and R. Stevenson, *Tetrahedron Letters*, 1966, 3063; A. J. Birch and G. S. R. Subba Rao, *ibid.*, 1967, 857; T. B. Windholz, B. Arison, R. D. Brown, and A. A. Patchett, *ibid.*, p. 3331.

<sup>2</sup> A. Chatterjee and B. G. Hazra, *Tetrahedron Letters*, 1969, 73.

<sup>3</sup> J. F. Grove, *J. Chem. Soc. (C)*, 1969, 549, and references cited therein.